

Published on Web 01/23/2009

Couplings between Peptide Linkages across a 3₁₀-Helical Hydrogen Bond Revealed by Two-Dimensional Infrared Spectroscopy

Hiroaki Maekawa,[†] Matteo De Poli,[‡] Claudio Toniolo,[‡] and Nien-Hui Ge*,[†]

Department of Chemistry, University of California at Irvine, Irvine, California 92697-2025, and Institute of Biomolecular Chemistry, CNR, Padova Unit, Department of Chemistry, University of Padova, 35131 Padova, Italy

Received September 23, 2008; E-mail: nhge@uci.edu

Peptide linkages connected through different types of hydrogen bonding networks are fundamental to various protein secondary structures. For example, the 3₁₀-helix exhibits intramolecular C=O··· H-N hydrogen bond pattern between the *i*th and (*i* + 3)th residues, whereas the α -helix is between the *i*th and (*i* + 4)th.¹ Vibrational couplings between peptide linkages are sensitive to the underlying structure, a feature extensively explored in IR experiments of amide-I modes to determine secondary structures.^{2,3} Here we demonstrate that vibrational coupling between hydrogen-bonded peptide linkages in a 3₁₀-helix can be directly probed by a combination of two-dimensional infrared (2D IR) spectroscopy and isotope editing of amide-I and amide-II modes. The isotope-dependent cross-peaks provide valuable information on local structure, such as the formation of a single helix turn, important for studying protein folding processes.

Recently, we have demonstrated that 2D IR of amide-I modes is a powerful technique that can distinguish the subtle conformational differences between 3_{10} - and α -helices as well as probe the onset of 310-helical secondary structure.⁴ The doublet amide-I cross-peak pattern is characteristic of the 310-helix structure, but contains no site-specific information. To elucidate the 310-helix formation at the residue level, we need to devise a strategy to directly probe the coupled vibrational modes that are connected through a 310-helical C=O····H-N hydrogen bond and involve significant motions of these atoms. Isotope substitution with ${}^{13}C={}^{16}O$ and ${}^{13}C={}^{18}O$ proves to be useful for isolating local amide-I modes and probing local structure and dynamics.⁵ Isotope labeling the hydrogen bonding partner with N-D is the most intuitive choice, but cannot be achieved in a site specific manner. The amide-II mode, consisting mainly of the N-H bending and C-N stretching,² is chosen as our reporter. ¹⁵N labeling decreases the amide-II frequency of N-methylacetamide (NMA) in nitrogen matrix by $13-15 \text{ cm}^{-1}$,⁶ a sizable shift compared to typical line widths (~30 cm⁻¹). Most 2D IR studies of peptides have focused on the amide-I mode.^{3–5} Only a recent few studies explored the couplings between the amide-I and amide-II modes in NMA and dipeptides,^{7,8} but cross-peaks have yet to be observed for modes connected through an intramolecular hydrogen bond in peptides.

We performed linear and 2D IR experiments on three hexapeptides in CDCl₃ (Figure 1a): Z-Aib-L-Leu-(Aib)₂-Gly-Aib-OtBu (1, Z, benzyloxycarbonyl; Aib, α -aminoisobutyric acid; OtBu, *tert*-butoxy), monolabeled peptide with ¹³C=¹⁸O at Leu (1*), and bis-labeled peptide with ¹³C=¹⁸O at Leu and ¹⁵N at Gly (1**). The 2D IR cross-peak pattern of 1 in CDCl₃ exhibits a clear doublet in the amide-I region (Figure S2), indicating that the peptide forms a 3₁₀-helix. This structure is not unexpected because many peptides with a high fraction of Aib are known to adopt the 3₁₀-helical conformation.⁹

Figure 1b shows the linear IR spectra of hexapeptides in the region covering the ${}^{13}C={}^{18}O$ amide-I mode and the amide-II modes. Two



Figure 1. (a) Molecular structure of 1 taking an ideal 3_{10} -helical conformation with the dihedral angles $(\phi, \psi) = (-57^{\circ}, -30^{\circ})^{.1}$ The atoms labeled with ¹³C, ¹⁸O, and ¹⁵N for 1* and 1** are colored green. (b) Normalized FTIR spectra of 1 (black), 1* (red), and 1** (blue) in CDCl₃. (c) Difference spectra in the amide-II region.

amide-II bands are observed at 1498 and 1528 cm⁻¹. Roughly speaking, the higher (lower) frequency band can be attributed to the vibrational exciton band of the hydrogen-bonded (free) amide-II modes. The amide-II mode is known to exhibit a blue shift when the N–H group is hydrogen bonded.¹⁰ If the hexapeptide forms a full 3_{10} -helix, four out of the five amide N–H groups will participate in intramolecular hydrogen bonding except for the first one at the N-terminus (Figure 1a). The relative intensity of the two bands in Figure 1b is consistent with this picture when compared to the trend observed in FTIR measurements of Z-(Aib)_n-OtBu (n = 1, 2, 3, 5, 8, and 10) in CDCl₃, where the intensity of the higher frequency band increases with *n* (Figure S3). The spectrum of Z-Aib-OtBu (Figure S1) shows that the capping groups provide only small contributions to the spectral window of interest.

The ¹³C=¹⁸O labeling on the Leu residue of **1*** results in an amide-I peak at 1598 cm⁻¹, completely separated from the much stronger ¹²C=¹⁶O amide-I bands (Figures 1b and S1). The isotopes also affect the amide-II modes. As shown in the difference spectrum (Figure 1c, **1***-**1**), the intensity at 1536 cm⁻¹ decreases and at 1514 cm⁻¹ increases. Such a red shift of the higher frequency band indicates that the N-H group of the second peptide linkage is hydrogen-bonded with the urethane C=O group, as expected for a full 3₁₀-helix (Figure 1a).

The bis-labeled peptide 1** has almost the same amide-I bands as 1*, so ¹⁵N labeling primarily affects the amide-II modes. The higher frequency band of 1** is red-shifted by ~4 cm⁻¹ from that of 1*. The difference spectrum between 1** and 1* reveals clearly the additional effect of ¹⁵N labeling, with major spectral changes occurring in the high frequency band. This indicates that the N–H group of the fourth peptide linkage is involved in hydrogen bonding. For a 3₁₀-helix, its bonding partner is the C=O group of the second peptide linkage.

The isotope effects on amide-II bands in Figure 1c are quite complex because the isotope shifts are small. Our DFT calculations on NMA in the gas phase at the B3LYP/6-311++G(d,p) level¹¹ predict 7 and 13 cm⁻¹ red shifts for ¹³C=¹⁸O and ¹⁵N substitution, respectively. The labeled amide-II local mode is still coupled to the network of unlabeled amide-II local modes. The labels on hydrogen-bonded linkages slightly affect the low frequency exciton band (see the minor intensity changes

[†] University of California at Irvine.

^{*} University of Padova.



Figure 2. Normalized absolute magnitude 2D IR rephasing (R, a-c) and nonrephasing (NR, d-f) spectra of **1**, **1***, and **1**** (from left to right) in CDCl₃. Horizontal slices of the NR spectra along the dashed line at $\omega_{\tau} = 1598 \text{ cm}^{-1}$ are shown in the top panels.

at 1492 cm^{-1} in Figure 1c), indicating that the free and hydrogenbonded amide-II local modes are coupled.

Figure 2 presents 2D IR rephasing (R) and nonrephasing (NR) spectra. Overall, we observed better resolved diagonal peaks and more distinct cross-peaks in the NR spectra than in the R spectra. The NR sequence has higher resolving power due to destructive interference effects.8 It is also very useful for line-narrowing cross-peaks between anticorrelated vibrators,12 such as hydrogen-bonded amide-I and amide-II modes. In the R spectra of 1^* and 1^{**} , the ¹³C=¹⁸O amide-I mode at the second peptide linkage is observed at $(\omega_t, \omega_\tau) \approx (1593, -1598)$ cm^{-1} and no peak appears here in the spectrum of **1**. For the amide-II region, the strongest diagonal peak at (1525, -1528) cm⁻¹ in the R spectrum of 1 corresponds to the high frequency band in the linear spectrum. Upon isotope labeling, the amide-II band maximum maintains at about the same position for 1^* , but shifts to (1520, -1521) cm^{-1} for 1**. The free amide-II band appears in the 2D R spectrum of 1 as a diagonally elongated shoulder on the red side of the main peak, but it is clearly resolved as a separate peak in the NR spectrum at (1493, 1497) cm⁻¹. The complex couplings among the amide-II modes manifest as many cross-peaks in the NR spectra.

The most interesting feature in the 2D spectra is the cross-peaks between the ${}^{13}C={}^{18}O$ amide-I mode and the amide-II modes. They are weak in the R spectrum, but much stronger in the NR spectrum. For **1***, the cross-peaks are located at (1527, 1598) and (1596, 1524) cm⁻¹. These cross-peaks reveal the presence of couplings between the labeled amide-I mode and some modes within the high frequency amide-II band. The couplings can involve anharmonic interactions between the second amide-I and amide-II on other linkage and/or between the second amide-I and amide-II on other linkages. Whether the fourth linkage contributes to the latter interactions can be revealed by observing the changes of these cross-peaks upon ¹⁵N labeling.

The top panels in Figure 2 show the slices through 2D NR spectra at $\omega_{\tau} = 1598 \text{ cm}^{-1}$. The cross-peak maximum of 1^{**} is at 1518 cm^{-1} , significantly red-shifted from that of 1^* by 9 cm⁻¹. This is a clear evidence that the amide-II local mode on the fourth peptide linkage is coupled to the amide-I local mode on the second peptide unit. ¹⁵N labeling changes the fourth amide-II local mode frequency and results in corresponding shifts in the cross-peak frequencies. Such coupling information is not obtainable from linear IR. The cross-peak line shapes observed here are quite complex because there are many amide-II exciton states that the ¹³C=¹⁸O labeled amide-I mode can couple to. Although detailed analysis is still underway, it may be worthwhile to estimate how this coupling depends on the existence of 3₁₀-helix based

on the transition charge coupling model.¹³ Using partial charges and charge derivatives along the amide-I and -II modes of NMA from a DFT calculation, we estimated the coupling between the two modes separated by two peptide units to be -12 cm^{-1} in the ideal 3₁₀-helix $[(\phi, \psi) = (-57^\circ, -30^\circ)]^1$ and -0.1 cm^{-1} in semiextended structure $[(-78^\circ, 146^\circ)]$. Such a difference in the coupling strength suggests that these cross-peaks would be useful for detecting helix formation. In a recent experiment the couplings between labeled amide-I modes separated by 1-3 residues have been measured in an alanine rich α -helix.^{5a} Measuring the amide-I/II couplings with labels at several different sites will enable further refinement of the peptide local structure. Also it would be interesting to theoretically investigate the amide-I/II coupling mechanism, including through-(hydrogen)-bond and through-space interactions. Although our new observation for the coupled amide-I and amide-II modes was made in a 310-helix, the technique is generally applicable to systems involving couplings between C=O stretching and N-H bending modes.

In summary, the combination of multiple isotope labels and 2D IR enabled us to observe the cross-peaks between the amide-I and amide-II modes connected through a C=O···H-N intramolecular hydrogen bond in a 3_{10} -helical peptide. Although the ¹⁵N isotope effect on the amide-II band is not as strong as ¹³C and ¹⁸O on the amide-I band, the 2D IR cross-peaks exhibit a clearly discernible ¹⁵N shift, twice that of linear IR. Thus, the ¹⁵N label is potentially useful for probing local conformations by 2D IR spectroscopy. The strategy demonstrated here is promising for directly detecting the transient formation of a single helix turn, which would be difficult to obtain from 2D IR of unlabeled amide-I modes.

Acknowledgment. This work was supported by grants from the American Chemical Society Petroleum Research Fund (39148-G6) and the National Science Foundation (CHE-0450045) to N.-H.G.

Supporting Information Available: Complete ref 11. Hexapeptide synthesis; measurement procedure; double-crossed polarization 2D IR spectrum of 1; and linear spectra of hexapeptides and Z-(Aib)_n-OtBu. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Toniolo, C.; Benedetti, E. Trends Biochem. Sci. 1991, 16, 350.
- (2) Krimm, S.; Bandekar, J. Adv. Protein Chem. 1986, 38, 181.
- (a) Hamm, P.; Helbing, J.; Bredenbeck, J. Annu. Rev. Phys. Chem. 2008, 59, 291.
 (b) Ganim, Z.; Chung, H. S.; Smith, A. W.; DeFlores, L. P.; Jones, K. C.; Tokmakoff, A. Acc. Chem. Res. 2008, 41, 432.
- (4) (a) Maekawa, H.; Toniolo, C.; Moretto, A.; Broxterman, Q. B.; Ge, N.-H. J. Phys. Chem. B 2006, 110, 5834. (b) Maekawa, H.; Toniolo, C.; Broxterman, Q. B.; Ge, N.-H. J. Phys. Chem. B 2007, 111, 3222. (c) Maekawa, H.; Formaggio, F.; Toniolo, C.; Ge, N.-H. J. Am. Chem. Soc. 2008, 130, 6556.
- (5) (a) Fang, C.; Wang, J.; Kim, Y. S.; Charnley, A. K.; Barber-Armstrong, W.; Smith, A. B., III; Decatur, S. M.; Hochstrasser, R. M. J. Phys. Chem. B 2004, 108, 10415. (b) Mukherjee, P.; Kass, I.; Arkin, I. T.; Zanni, M. T. Proc. Natl. Acad. Sci. U.S.A. 2006, 103, 3528. (c) Smith, W. S.; Tokmakoff, A. J. Chem. Phys. 2007, 126, 045109. (d) Fang, C.; Senes, A.; Cristian, L.; DeGrado, W. F.; Hochstrasser, R. M. Proc. Natl. Acad. Sci. U.S.A. 2006, 103, 16740.
- (6) Fillaux, F.; De Lozé, C. J. Chim. Phys. Phys.-Chim. Biol. 1976, 73, 1004.
 (7) (a) Rubtsov, I. V.; Wang, J.; Hochstrasser, R. M. Proc. Natl. Acad. Sci. U.S.A. 2003, 100, 5601. (b) DeFlores, L. P.; Ganim, Z.; Ackley, S. F.;
- U.S.A. 2005, 100, 5001. (b) DePlores, L. P.; Ganim, Z.; Ackley, S. F.; Chung, H. S.; Tokmakoff, A. J. Phys. Chem. B 2006, 110, 18973. (c) DeFlores, L. P.; Tokmakoff, A. J. Am. Chem. Soc. 2006, 128, 16520. (d) Hayashi, T.; Mukamel, S. J. Phys. Chem. B 2007, 111, 11032.
 Sul, S.; Karajskai, D.; Jiang, Y.; Ge, N. H. J. Phys. Chem. B 2006, 110.
- (8) Sul, S.; Karaiskaj, D.; Jiang, Y.; Ge, N. H. J. Phys. Chem. B 2006, 110, 19891.
- (9) Karle, I. L.; Balaram, P. *Biochemistry* **1990**, *29*, 6747.
- (10) Kim, J.-H.; Cho, M. Bull. Korean Chem. Soc. 2003, 24, 1061.
- (11) Frisch, M. J.; et al. *Gaussian 03*; Gaussian, Inc.: Wallingford, CT, 2004.
 (12) Ge, N.-H.; Zanni, M. T.; Hochstrasser, R. M. *J. Phys. Chem. A* 2002, *106*, 962
- (13) Hamm, P.; Hochstrasser, R. M. In Ultrafast Infrared and Raman Spectroscopy; Fayer, M. D., Ed.; Marcel Dekker: New York, 2001; pp 273–347.
- JA807572F